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Synthesis of novel 1,4,5-trisubstituted 3-trifluoromethylpyrazoles via microwave-assisted Stille coupling reactions

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Abstract

1,5-Disubstituted 3-trifluoromethylpyrazoles were reacted with N-bromosuccinimide in DMF at room temperature or 70-80 °C for 1-2 h to afford the corresponding 4-bromo-substituted pyrazoles 2 in 95–99% yields. The microwave-assisted Stille coupling reactions of 2 with arylstannanes having a substituent on the benzene ring and allylstannane in refluxing CH₃CN in the presence of Pd(PPh₃)₄ provided the corresponding 1,4,5-trisubstituted 3-trifluoromethylpyrazoles 3 in 75–98% yields. © 2007 Elsevier B.V. All rights reserved.

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1. Introduction

The pyrazole ring has been known as an important framework in a large number of compounds possessing pharmaceutical and agrochemical properties [1-6]. Numerous methods for the synthesis of this family of compounds have been well documented in the previous literatures [7-15], but methods are mostly related to the synthesis of nonfluorinated pyrazoles. However, to our knowledge, there have been limited methodologies for the preparation of trifluoromethylated pyrazoles. Since the introduction of a trifluoromethyl group into the pyrazole ring system has often increased the biological properties [16,17], many efforts have been devoted to the development of synthetic methodologies for the preparation of trifluoromethylated pyrazoles. General methods for the preparation of these compounds involve the reactions of hydrazine derivatives with trifluoromethylated precursors such as 1-trifluoromethylated 1.3-diketones [16,18,19], trifluoromethylacetylenic esters [20], pentafluoroethylacetylenes [21], trifluoroacetyl acetylenes [22], β-alkoxyvinyl trifluoromethyl ketones [23,24], β-trifluoromethyl enaminones [25,26], N-aryl-

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1-trifluoromethyl acetylenic imines [27] and 1-pentafluoroethyl-2-iodoalkenes [28,29]. 1,3-Dipolar cycloaddition reaction of 2-bromo-3,3,3-trifluoropropene with diazomethane followed by dehydration also afforded the corresponding 3trifluoromethylpyrazole [30]. Recently, a direct preparation of 5-tributylstannyl-4-trifluoromethylpyrazole was accomplished by 1,3-dipolar cycloaddition reaction of tributyl(3,3,3-trifluoro-1-propynyl)stannane with diazomethane [31,32]. Dipolarophiles such as phenylacetylene, vinylsulfone and 2-bromoacrylate were also reacted with trifluoroacetyltriphenylsilane 2,4,6-triisopropylbenzenesulfonyl hydrazone to give the corresponding 3-trifluoromethylpyrazoles [33]. However, the most of these methods focused on the preparation of mono- and disubstituted trifluoromethylpyrazole derivatives. In spite of importance of trisubstituted trifluoromethylpyrazole in recent years, the methods for the preparation of trisubstituted trifluoromethylpyrazoles have been quite limited in the previous literatures [16,34–36]. Moreover, these methods have some drawbacks such as a lack of generality and uneasy availability of starting materials. Herein, we wish to report a novel and general approach to 1,4,5-trisubstituted 3-trifluoromethylpyrazoles via the microwave-assisted Stille coupling reactions of the corresponding 1,5-disubstituted 4-bromo-3trifluoromethylpyrazoles with arylstannanes or allylstannane in the presence of palladium catalyst.

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2. Results and discussion

The starting materials, 1,5-disubstituted 3-trifluoromethylpyrazoles 1a-f, were prepared in 92-98% isolated yields from the condensation reaction of 1-trifluoromethylated 1,3-dicarbonyl compounds with hydrazine derivatives in 1,4-dioxane, except for methylhydrazine, followed by dehydration reaction in refluxing toluene. Small quantities of undesired regioisomer, 1,3-disubstituted 5-trifluoromethylpyrazoles, were separated by column chromatographic purification. A recent study has established that this high regioselectivity was due to electronwithdrawing group such as CF₃ on C-1 of 1,3-dicarbonyl compounds [37]. In order to install the aryl or allyl substitutent onto the 4-position of fully assembled pyrazoles 1, we chose to explore a carbon-carbon bond-forming reaction in the presence of transition-metal catalysis. Therefore, we carried out the bromination reaction with the corresponding bromo succinimide to introduce a bromo substituent as a reacting site of the coupling reaction. We found 1,5-disubstituted 4-bromo-3trifluoromethylpyrazoles 2a-g were prepared in 95-99% isolated yields from the reaction of 1 with N-bromosuccinimide in DMF at room temperature or 70-80 °C. The results of these reactions were summarized in Table 1. The use of other solvents such as CH₂Cl₂, CCl₄ or CH₃CN in this bromination reaction caused not only to retard the reaction, but also to decrease the yield of 2. The previous method using sodium acetate and bromine did not provide a satisfied result for the formation of 2 [38].

Since 5-substituted 3-trifluoromethyl-1-methylpyrazoles were not directly synthesized from the reaction of 1-trifluoromethylated 1,3-dicarbonyl compounds with methylhydrizine, 5-substituted 4-bromo-3-trifluoromethyl-1-methylpyrazoles 2h-j were prepared regiospecifically from the methylation reaction of 2a, 2d and 2g with methyl iodide in the presence of K₂CO₃ (Scheme 1). The assignment of regioisomers 2h-j was based on ¹H NMR chemical shifts. The ¹H NMR spectrum of 2h-j showed that the peaks of *N*-methyl protons appeared at $\delta = 3.84$ -3.95 ppm which is consistent with the range of chemical shift of a similar structure, 4-substituted

Table 1



i,J-uisubsiitu	icu 4-bioino-	-5-umuorom	emyipyiaz			
$ \begin{array}{c} CF_{3} \\ N \\ N \\ 2 \end{array} \xrightarrow{\text{CF}_{3}} \\ DMF, T (^{\circ}C), t (h) \\ R^{1} \\ R^{2} \end{array} \xrightarrow{\text{CF}_{3}} \\ R^{1} \\ N \\ R^{2} \end{array} $						
		2				
R^1	\mathbb{R}^2	<i>T</i> (°C)	<i>t</i> (h)	Yield (%) ^a		
Ph	Н	70-80	2	96		
Ph	PhCH ₂	70-80	2	98		
Ph	Ph	70-80	2	99		
Me	Н	25	1	99		
Me	$PhCH_2$	25	1	95		
cy-Pr	Ph	25	1	98		
cy-Pr	Н	25	1	99		
	NBS DMF, T (°C R ¹ Ph Ph Ph Me Me cy-Pr cy-Pr	NBS DMF, T ($^{\circ}$ C), t (h) R ¹ R ¹ R ² Ph H Ph PhCH ₂ Ph Ph Me H Me PhCH ₂ cy-Pr Ph cy-Pr H	$\begin{array}{c c} & & & & \\ \hline \hline & & & \\ \hline & & & \\ \hline \hline \\ \hline & & & \\ \hline \end{array}$	$\begin{array}{c c} & & \\ \hline NBS \\ \hline DMF, T (^{\circ}C), t (h) \end{array} \xrightarrow{R^{1}} R^{1} \\ \hline R^{2} \\ \hline R^{1} \\ \hline R^{2} \\ $		

^a lsolated yield.



3-trifluoromethyl-1-methylpyrazoles [39]. The resonance of *N*-methyl protons of regioisomer, 4-substituted 5-trifluoromethyl-1-methylpyrazoles, appeared at $\delta = 4.07-4.11$ ppm. Regiospecificity on *N*-methylation of **2a**, **2d** and **2g** to give 5-substituted 4-bromo-3-trifluoromethyl-1-methylpyrazole **2h–j** can be explained by the steric effect of trifluoromethyl group in the formation of regioisomer, 3-substituted 4-bromo-5-trifluoromethyl-1-methylpyrazole. A similar result was established in the formation of 1-methyl-3-trimethylsilylpyrazole and 1-methyl-3,4-bis(trimethylsilyl)pyrazole [40].

Although several previous literature reports described the coupling reactions, such as Suzuki, Stille, and Sonogashira couplings, performed on nonfluorinated 5-bromopyrazoles [10,41] and pyrazole triflates [42], there has been only one example (Sonogashira reaction) in the coupling reactions on trifluoromethylated halopyrazoles [38]. Among the coupling reactions, we investigated Stille coupling reaction of 2 with arylstannane or allylstannane. We carried out the crosscoupling reaction of 2c with phenylstannane as a standard reaction to find out an optimized reaction condition. When 2c was reacted with phenylstannane in the presence of $Pd(PPh_3)_4$ (5 mol%) at refluxing CH₃CN for 24 h, the cross-coupled product 3a was obtained in 76% isolated yield based on the only 15% conversion of the starting material. The use of DMF as a solvent under the same reaction condition caused to enhance the conversion of the starting material up to 40%, but reaction was still sluggish. Recently, we reported that microwaveassisted cross-coupling reaction of trifluoromethylated chloropyrimidines with arylstannanes afforded the corresponding coupled product in high yields [43]. We applied this reaction condition to explore the coupling reaction of 2 with arylstannane or allylstannane. It was found that microwaveassisted coupling reaction of 2c with phenylstannane in CH₃CN at 180 °C for 1 h afforded the coupled product 3a in 97% isolated yield based on the 100% conversion of 2c. The arylstannane having a substituent on the benzene ring and allylstannane also underwent the coupling reactions with 2c to give **3b-e** in 92–98% isolated yields under the same reaction condition. The reactions between 2b and arylstannane also afforded the corresponding coupled products 3f-i in 91-92% isolated yields, but the reactions were more sluggish than that of 2c with phenylstannane and a prolong reaction time (1.5 h) was required. A small amount of reduced product was obtained in less than 5% yield for each case. The reaction of 2b with allylstannane under the same reaction condition afforded the corresponding coupled product 3j in 97% isolated yield. Similarly, 2g underwent the coupling reaction with Table 2

The coupling reactions of 1,5-disubstituted 4-bromo-3-trifluoromethylpyrazoles **2** with arylstannanes or allystannane



Compound no.	R^1	R^2	R	<i>t</i> (h)	Yield (%) ^a
3a	Ph	Ph	C ₆ H ₅	1	97
3b	Ph	Ph	p-MeO-C ₆ H ₄	1	96
3c	Ph	Ph	p-F-C ₆ H ₄	1	98
3d	Ph	Ph	p-CF ₃ -C ₆ H ₄	1	92
3e	Ph	Ph	CH2=CH-CH2	1	97
3f	Ph	Bn	C ₆ H ₅	1.5	92
3g	Ph	Bn	p-MeO-C ₆ H ₄	1.5	92
3h	Ph	Bn	p-F-C ₆ H ₄	1.5	91
3i	Ph	Bn	p-CF ₃ -C ₆ H ₄	1.5	92
3ј	Ph	Bn	CH2=CH-CH2	1.5	97
3k	Ph	Me	C ₆ H ₅	1.5	81
31	Ph	Me	p-MeO-C ₆ H ₄	1.5	82
3m	Ph	Me	p-F–C ₆ H ₄	1.5	84
3n	Ph	Me	p-CF ₃ -C ₆ H ₄	1.5	82
30	Ph	Me	CH2=CH-CH2	1.5	90
3p	cy-Pr	Ph	C ₆ H ₅	2.5	91
3q	cy-Pr	Ph	CH2=CH-CH2	2.5	93
3r	cy-Pr	Me	C ₆ H ₅	2.5	75
3s	cy-Pr	Me	CH2=CH-CH2	2.5	80

^a lsolated yield.

arylstannanes or allylstannane to give the coupled products 3k-o in 81-90% isolated yields. The coupling reactions of 5-cyclopropyl substituted pyrazole derivative 2f with phenylstannane or allylstannane were much sluggish under the same reaction condition and a prolong reaction time (2.5 h) was required. The results of these coupling reactions were summarized in Table 2.

3. Conclusion

In conclusion, we synthesized 1,4,5-trisubstituted 3-trifluoromethylpyrazoles in excellent yields via the microwave-assisted Stille coupling reactions of 1,5-disubstituted 4-bromo-3-trifluoromethylpyrazoles with arylstannane or allylstannane in the presence of Pd(PPh₃)₄ catalyst. 1,5-Disubstituted 4-bromo-3trifluoromethylpyrazoles were easily prepared by bromination of 1,5-disubstituted 3-trifluoromethylpyrazoles with NBS. The present method provided an efficient and facile approach to novel 1,4,5-trisubstituted 3-trifluoromethylpyrazoles.

4. Experimental

¹H NMR spectra were recorded on a 200 MHz Gemini-200 NMR spectrometer and ¹⁹F NMR spectra were recorded on a 100 MHz Bruker AC-100F NMR spectrometer with tetramethylsilane (TMS) and CFCl₃ as an internal standard, respectively and the upfield as negative. All chemical shifts (δ) are expressed in parts per million and coupling constant (*J*) are given in Hertz. Infrared spectra were determined on a Mattson Genesis series FT High Resolution Spectrophotometer. Mass spectra were obtained by using GC/MS-Qp1000-Shimadzu (EI, 70 eV). Microwave-assisted coupling reactions were carried by using Biotage Microwave Organic Synthesizer. Melting points were determined in open capillary tubes and are uncorrected.

Commercially available reagents were purchased from Aldrich, Lancaster, Tokyo Kasei and Fluorochem. All solvents were dried by general purification method. Flash chromatography was performed on 40–60 μ m silica gel (230–400 mesh).

4.1. General procedure for the preparation of 4brominated pyrazoles 2a-2g

A 50 mL two-neck round bottom flask equipped with a magnetic stirrer bar, a septum and reflux condenser connected to an argon source was charged with 3-trifluoromethylated pyrazoles 1 (2.0 mmol), NBS (4.0 mmol) and 10 mL of DMF. After the reaction mixture was stirred at room temperature for 1 h or heated at 70–80 °C for 2 h, the reaction mixture was quenched with water. The reaction mixture was extracted with methylene chloride twice, dried over anhydrous MgSO₄ and chromatographed on SiO₂ column. Elution with a mixture of hexane and ethyl acetate (9:1) provided 4-brominated pyrazoles **2a–2g**.

4.1.1. 4-Bromo-3-trifluoromethyl-5-phenylpyrazole (2a)

2a was prepared in 96% yield (0.559 g) according to the general procedure (the reaction mixture was heated at 70–80 °C for 2 h). **2a**: mp 173–175 °C; ¹H NMR (CDCl₃) δ 11.06 (bs, 1H), 7.69–7.50 (m, 5H); ¹⁹F NMR (CDCl₃, internal standard CFCl₃) δ –63.22 (s, 3F); MS, *m/z* (relative intensity) 292 (*M*⁺ + 1, 52), 290 (*M*⁺ – 1, 53), 182 (12), 163 (31), 142 (100), 133 (35), 89 (19), 77 (30). Anal. Calcd for C₁₀H₆BrF₃N₂: C, 41.26; H, 2.08. Found: C, 41.10; H, 2.05.

4.1.2. 1-Benzyl-4-bromo-3-trifluoromethyl-5phenylpyrazole (2b)

2b was prepared in 98% yield (0.747 g) according to the general procedure (the reaction mixture was heated at 70–80 °C for 2 h). **2b**: oil; ¹H NMR (CDCl₃) δ 7.50–7.44 (m, 3H), 7.27–7.23 (m, 5H), 7.00–6.97 (m, 2H), 5.28 (s, 2H); ¹⁹F NMR (CDCl₃, internal standard CFCl₃) δ –62.65 (s, 3F); MS, *m/z* (relative intensity) 382 (*M*⁺ + 1, 10), 380 (*M*⁺ – 1, 11), 91 (100), 65 (17). Anal. Calcd for C₁₇H₁₂BrF₃N₂: C, 53.57; H, 3.17. Found: C, 53.43; H, 3.12.

4.1.3. 4-Bromo-3-trifluoromethyl-1,5-diphenylpyrazole (*2c*)

2c was prepared in 99% yield (0.727 g) according to the general procedure (the reaction mixture was heated at 70–80 °C for 2 h). **2c**: mp 102–104 °C; ¹H NMR (CDCl₃) δ 7.43–7.20 (m, 10H); ¹⁹F NMR (CDCl₃, internal standard CFCl₃) δ –62.92 (s, 3F); MS, *m*/*z* (relative intensity) 368 (*M*⁺ + 1, 64), 366 (*M*⁺ – 1, 66), 347 (24), 286 (23), 218 (13), 120 (24), 89 (23), 77 (100). Anal. Calcd for C₁₆H₁₀BrF₃N₂: C, 52.34; H, 2.75. Found: C, 52.11; H, 2.70.

4.1.4. 4-Bromo-3-trifluoromethyl-5-methylpyrazole (2d)

2d was prepared in 99% yield (0.453 g) according to the general procedure (the reaction mixture was stirred at room temperature for 1 h). **2d**: mp 89–90 °C; ¹H NMR (CDCl₃) δ 11.88 (bs, 1H), 2.33 (s, 3H); ¹⁹F NMR (CDCl₃, internal standard CFCl₃) δ –63.15 (s, 3F); MS, *m/z* (relative intensity) 230 (*M*⁺ + 1, 14), 228 (*M*⁺ – 1, 15), 211 (4), 149 (19), 101 (69), 69 (100). Anal. Calcd for C₅H₄BrF₃N₂: C, 26.22; H, 1.76. Found: C, 26.18; H, 1.74.

4.1.5. 1-Benzyl-4-bromo-3-trifluoromethyl-5methylpyrazole (2e)

2e was prepared in 95% yield (0.606 g) according to the general procedure (the reaction mixture was stirred at room temperature for 1 h). **2e**: oil; ¹H NMR (CDCl₃) δ 7.38–7.30 (m, 3H), 7.15–7.11 (m, 2H), 5.34 (s, 2H), 2.19 (s, 3H); ¹⁹F NMR (CDCl₃, internal standard CFCl₃) δ –63.29 (s, 3F); MS, *m*/*z* (relative intensity) 320 (*M*⁺ + 1, 14), 318 (*M*⁺ – 1, 13), 305 (10), 249 (7), 149 (5), 91 (100), 65 (19). Anal. Calcd for C₁₂H₁₀BrF₃N₂: C, 45.16; H, 3.16. Found: C, 44.99; H, 3.11.

4.1.6. 4-Bromo-5-cyclopropyl-3-trifluoromethyl-1phenylpyrazole (2f)

2f was prepared in 98% yield (0.649 g) according to the general procedure (the reaction mixture was stirred at room temperature for 1 h). **2f**: oil; ¹H NMR (CDCl₃) δ 7.53–7.43 (m, 5H), 1.80–1.74 (m, 1H), 0.96–0.88 (m, 2H), 0.75–0.69 (m, 2H); ¹⁹F NMR (CDCl₃, internal standard CFCl₃) δ –63.18 (s, 3F); MS, *m*/*z* (relative intensity) 332 (*M*⁺ + 1, 8), 330 (*M*⁺ – 1, 9), 295 (10), 236 (16), 128 (13), 77 (100). Anal. Calcd for C₁₃H₁₀BrF₃N₂: C, 47.15; H, 3.04. Found: C, 47.42; H, 2.99.

4.1.7. 4-Bromo-5-cyclopropyl-3-trifluoromethylpyrazole (*2g*)

2g was prepared in 99% yield (0.505 g) according to the general procedure (the reaction mixture was stirred at room temperature for 1 h). **2g**: mp 112–114 °C; ¹H NMR (CDCl₃) δ 11.88 (bs, 1H), 1.90–1.76 (m, 1H), 1.16–0.99 (m, 2H), 0.84–0.76 (m, 2H); ¹⁹F NMR (CDCl₃, internal standard CFCl₃) δ –63.16 (s, 3F); MS, *m/z* (relative intensity) 256 (*M*⁺ + 1, 17), 354 (*M*⁺ – 1, 18), 155 (18), 105 (22), 69 (23), 40 (100). Anal. Calcd for C₁₇H₆BrF₃N₂: C, 32.97; H, 2.37. Found: C, 32.85; H, 2.34.

4.2. General procedure for the preparation of 4brominated pyrazoles **2h**-2j

A 50 mL two-neck round bottom flask equipped with a magnetic stirrer bar, a septum and reflux condenser connected to an argon source was charged with 3-trifluoromethylated 4-bromopyrazoles **2a**, **2d** or **2g** (1.5 mmol), K₂CO₃ (1.7 mmol) and 10 mL of dry Acetone. After the reaction mixture was stirred at room temperature for 1 h, CH₃I (2.0 mmol) was added into the reaction and then the reaction mixture was stirred at room temperature for 1.5 h. After removal of solvent, water was added. The reaction mixture was extracted with ethyl acetate twice, dried over anhydrous MgSO₄ and chromatographed on

 SiO_2 column. Elution with a mixture of hexane and ethyl acetate (9:1) provided 4-brominated pyrazoles **2h–2j**.

4.2.1. 4-Bromo-3-trifluoromethyl-1-methyl-5-

phenylpyrazole (**2h**)

2h was prepared in 60% yield (0.275 g) according to the general procedure. **2h**: mp 66–67 °C; ¹H NMR (CDCl₃) δ 7.56–7.51 (m, 3H), 7.49–7.36 (m, 2H), 3.84 (s, 3H); ¹⁹F NMR (CDCl₃, internal standard CFCl₃) δ –62.71 (s, 3F); MS, *m*/*z* (relative intensity) 306 (*M*⁺ + 1, 28), 304 (*M*⁺ – 1, 29), 223 (47), 153 (64), 127 (95), 104 (100), 77 (42). Anal. Calcd for C₁₁H₈BrF₃N₂: C, 43.30; H, 2.64. Found: C, 43.11; H, 2.69.

4.2.2. 4-Bromo-3-trifluoromethyl-1,5-dimethylpyrazole (2i)

2i was prepared in 79% yield (0.288 g) according to the general procedure. **2i**: mp 42–44 °C; ¹H NMR (CDCl₃) δ 3.85 (s, 3H), 2.27 (s, 3H); ¹⁹F NMR (CDCl₃, internal standard CFCl₃) δ –62.55 (s, 3F); MS, *m/z* (relative intensity) 244 (*M*⁺ + 1, 25), 242 (*M*⁺ – 1, 27), 223 (11), 163 (48), 99 (17), 69 (72), 56 (73), 43 (100). Anal. Calcd for C₆H₆BrF₃N₂: C, 29.65; H, 2.49. Found: C, 29.49; H, 2.51.

4.2.3. 4-Bromo-5-cyclopropyl-3-trifluoromethyl-1methylpyrazole (*2j*)

2j was prepared in 70% yield (0.377 g) according to the general procedure. **2j**: oil; ¹H NMR (CDCl₃) δ 3.94 (s, 3H), 1.72–1.57 (m, 1H), 1.15–1.05 (m, 2H), 0.99–0.86 (m, 2H); ¹⁹F NMR (CDCl₃, internal standard CFCl₃) δ –62.53 (s, 3F); MS, *m*/*z* (relative intensity) 270 (*M*⁺ + 1, 92), 268 (*M*⁺ – 1, 91), 169 (95), 91 (74), 58 (93), 44 (100). Anal. Calcd for C₈H₈BrF₃N₂: C, 35.71; H, 3.00. Found: C, 35.60; H, 2.95.

4.3. General procedure for the preparation of 1,4,5trisubstituted 3-trifluoromethylpyrazoles **3a–3s**

A 2 mL microwave reactor was charged with 2 (0.1 mmol), Pd(PPh₃)₄ (5 mol%) and 1.5 mL of dry CH₃CN. After bubbling with air gas, arylstannane or allylstannane (0.13 mmol) was added into the reactor. The reactor was heated at 180 °C for 1–2.5 h. After quenching with 10% KF solution at room temperature, the reaction mixture was extracted with CH₂Cl₂ twice. The CH₂Cl₂ solution was dried over anhydrous MgSO₄ and chromatographed on SiO₂ column. Elution with a mixture of hexane and ethyl acetate (9:1) provided **3a–3s**.

4.3.1. 3-Trifluoromethyl-1,4,5-triphenylpyrazole (3a)

3a was prepared in 97% yield (0.035 g) according to the general procedure (the reactor was heated at 180 °C for 1 h). **3a**: mp 159–161 °C; ¹H NMR (CDCl₃) δ 7.35–7.14 (m, 13H), 7.03–6.98 (m, 2H); ¹⁹F NMR (CDCl₃, internal standard CFCl₃) δ –59.98 (s, 3F); MS, *m/z* (relative intensity) 364 (*M*⁺, 22), 267 (10), 190 (12), 165 (12), 89 (11), 77 (100); IR (KBr) 3047, 2920, 1595, 1502, 1155, 1122, 755, 692 cm⁻¹. Anal. Calcd for C₂₂H₁₅F₃N₂: C, 72.52; H, 4.15. Found: C, 72.37; H, 4.09.

4.3.2. 3-Trifluoromethyl-4-(p-methoxy)phenyl-1,5diphenylpyrazole (**3b**)

3b was prepared in 96% yield (0.038 g) according to the general procedure (the reactor was heated at 180 °C for 1 h). **3b**: mp 124–125 °C; ¹H NMR (CDCl₃) δ 7.31–7.11 (m, 10H), 7.03–6.98 (m, 2H), 6.84–6.80 (m, 2H), 3.79 (s, 3H); ¹⁹F NMR (CDCl₃, internal standard CFCl₃) δ –60.05 (s, 3F); MS, *m/z* (relative intensity) 394 (*M*⁺, 90), 379 (16), 254 (10), 152 (13), 77 (100); IR (KBr) 3064, 2935, 1614, 1520, 1252, 1153, 1126, 825, 747, 697 cm⁻¹. Anal. Calcd for C₂₃H₁₇F₃N₂O: C, 70.04; H, 4.34. Found: C, 70.23; H, 4.27.

4.3.3. 3-Trifluoromethyl-4-(p-fluoro)phenyl-1,5diphenylpyrazole (**3c**)

3c was prepared in 98% yield (0.037 g) according to the general procedure (the reactor was heated at 180 °C for 1 h). **3c**: mp 161–162 °C; ¹H NMR (CDCl₃) δ 7.36–7.15 (m, 10H), 7.03–6.92 (m, 4H); ¹⁹F NMR (CDCl₃, internal standard CFCl₃) δ –60.06 (s, 3F), –114.90 (s, 1F); MS, *m/z* (relative intensity) 382 (*M*⁺, 25), 341 (11), 285 (9), 183 (9), 107 (10), 77 (100); IR (KBr) 3064, 2918, 1520, 1153, 1126, 817, 752, 691 cm⁻¹. Anal. Calcd for C₂₂H₁₄F₄N₂: C, 69.11; H, 3.69. Found: C, 69.33; H, 3.74.

4.3.4. 3-Trifluoromethyl-4-(p-trifluoromethyl)phenyl-1,5diphenylpyrazole (**3d**)

3d was prepared in 92% yield (0.040 g) according to the general procedure (the reactor was heated at 180 °C for 1 h). **3d**: mp 121–123 °C; ¹H NMR (CDCl₃) δ 7.57–7.52 (m, 2H), 7.38–7.17 (m, 10H), 7.03–6.97 (m, 2H); ¹⁹F NMR (CDCl₃, internal standard CFCl₃) δ –59.90 (s, 3F), –63.11 (s, 3F); MS, *m*/*z* (relative intensity) 432 (*M*⁺, 100), 391 (10), 363 (9), 267 (8), 146 (15), 77 (91); IR (KBr) 3060, 2924, 1600, 1487, 1327, 1169, 1128, 809, 750, 689 cm⁻¹. Anal. Calcd for C₂₃H₁₄F₆N₂: C, 63.89; H, 3.26. Found: C, 63.67; H, 3.20.

4.3.5. 4-Allyl-3-trifluoromethyl-1,5-diphenylpyrazole (3e)

3e was prepared in 97% yield (0.032 g) according to the general procedure (the reactor was heated at 180 °C for 1 h). **3e**: oil; ¹H NMR (CDCl₃) δ 7.40–7.16 (m, 10H), 6.01–5.82 (m, 1H), 5.08–4.91 (m, 2H), 3.29 (d, *J* = 5.9 Hz, 2H); ¹⁹F NMR (CDCl₃, internal standard CFCl₃) δ –61.51 (s, 3F); MS, *m/z* (relative intensity) 328 (*M*⁺, 53), 313 (42), 259 (15), 115 (22), 77 (100); IR (neat) 3064, 2922, 1641, 1599, 1493, 1277, 1165, 1128, 991, 915, 755, 694 cm⁻¹. Anal. Calcd for C₁₉H₁₅F₃N₂: C, 69.50; H, 4.60. Found: C, 69.27; H, 4.53.

4.3.6. 1-Benzyl-3-trifluoromethyl-4,5-diphenylpyrazole (3f)

3f was prepared in 92% yield (0.035 g) according to the general procedure (the reactor was heated at 180 °C for 1.5 h). **3f**: oil; ¹H NMR (CDCl₃) δ 7.43–7.01 (m, 15H), 5.30 (s, 2H); ¹⁹F NMR (CDCl₃, internal standard CFCl₃) δ –59.65 (s, 3F); MS, *m*/*z* (relative intensity) 378 (*M*⁺, 15), 301 (5), 189 (5), 91 (100), 77 (5); IR (neat) 3064, 2922, 1454, 1223, 1182, 1134, 753, 696 cm⁻¹. Anal. Calcd for C₂₃H₁₇F₃N₂: C, 73.01; H, 4.53. Found: C, 72.78; H, 4.59.

4.3.7. 1-Benzyl-3-trifluoromethyl-4-(p-methoxy)phenyl-5-phenylpyrazole (**3**g)

3g was prepared in 92% yield (0.038 g) according to the general procedure (the reactor was heated at 180 °C for 1.5 h). **3g**: mp 74–76 °C; ¹H NMR (CDCl₃) δ 7.35–7.24 (m, 6H), 7.09–7.02 (m, 6H), 6.78–6.74 (m, 2H), 5.30 (s, 2H), 3.75 (s, 3H); ¹⁹F NMR (CDCl₃, internal standard CFCl₃) δ –59.74 (s, 3F); MS, *m/z* (relative intensity) 408 (*M*⁺, 29), 339 (5), 176 (3), 151 (3), 91 (100); IR (KBr) 3032, 2920, 1520, 1250, 1178, 1134, 820, 749, 694 cm⁻¹. Anal. Calcd for C₂₄H₁₉F₃N₂O: C, 70.58; H, 4.69. Found: C, 70.39; H, 4.64.

4.3.8. 1-Benzyl-3-trifluoromethyl-4-(p-fluoro)phenyl-5-phenylpyrazole (3h)

3h was prepared in 91% yield (0.036 g) according to the general procedure (the reactor was heated at 180 °C for 1.5 h). **3h**: oil; ¹H NMR (CDCl₃) δ 7.41–6.85 (m, 14H), 5.30 (s, 2H); ¹⁹F NMR (CDCl₃, internal standard CFCl₃) δ –59.76 (s, 3F), –115.28 (s, 1F); MS, *m/z* (relative intensity) 396 (*M*⁺, 44), 377 (4), 319 (8), 207 (4), 103 (3), 91 (100); IR (neat) 3043, 2920, 1520, 1225, 1157, 1134, 825, 749, 688 cm⁻¹. Anal. Calcd for C₂₃H₁₆F₄N₂: C, 69.69; H, 4.07. Found: C, 69.48; H, 4.02.

4.3.9. 1-Benzyl-3-trifluoromethyl-4-(p-

trifluoromethyl) phenyl-5-phenylpyrazole~(3i)

3i was prepared in 92% yield (0.041 g) according to the general procedure (the reactor was heated at 180 °C for 1.5 h). **3i**: oil; ¹H NMR (CDCl₃) δ 7.50–7.46 (m, 2H), 7.40–7.24 (m, 8H), 7.08–7.04 (m, 4H), 5.30 (s, 2H); ¹⁹F NMR (CDCl₃, internal standard CFCl₃) δ –59.62 (s, 3F), –63.13 (s, 3F); MS, *m/z* (relative intensity) 446 (*M*⁺, 11), 377 (7), 189 (6), 91 (100), 65 (3); IR (neat) 3039, 2922, 1454, 1325, 1126, 828, 748, 685 cm⁻¹. Anal. Calcd for C₂₄H₁₆F₆N₂: C, 64.58; H, 3.61. Found: C, 64.27; H, 3.55.

4.3.10. 4-Allyl-1-benzyl-3-trifluoromethyl-5phenylpyrazole (**3***j*)

3j was prepared in 97% yield (0.033 g) according to the general procedure (the reactor was heated at 180 °C for 1.5 h). **3j**: mp 62–63 °C; ¹H NMR (CDCl₃) δ 7.45–7.34 (m, 3H), 7.27–7.13 (m, 5H), 6.98–6.94 (m, 2H), 5.91–5.71 (m, 1H), 5.22 (s, 2H), 4.94 (d, *J* = 10.3 Hz, 1H), 4.83 (d, *J* = 16.8 Hz, 1H), 3.17 (d, *J* = 5.9 Hz, 2H); ¹⁹F NMR (CDCl₃, internal standard CFCl₃) δ –61.27 (s, 3F); MS, *m/z* (relative intensity) 342 (*M*⁺, 3), 251 (5), 183 (4), 91 (100), 65 (26); IR (KBr) 3081, 2922, 1641, 1495, 1454, 1275, 1149, 1126, 989, 909, 752, 690 cm⁻¹. Anal. Calcd for C₂₀H₁₇F₃N₂: C, 70.17; H, 5.00. Found: C, 69.89; H, 4.93.

4.3.11. 3-Trifluoromethyl-1-methyl-4,5-diphenylpyrazole (*3k*)

3k was prepared in 81% yield (0.025 g) according to the general procedure (the reactor was heated at 180 °C for 1.5 h). **3k**: mp 108–110 °C; ¹H NMR (CDCl₃) δ 7.40–7.34 (m, 4H), 7.27–7.12 (m, 6H), 3.87 (s, 3H); ¹⁹F NMR (CDCl₃, internal standard CFCl₃) δ –59.76 (s, 3F); MS, *m/z* (relative intensity) 302 (*M*⁺, 100), 287 (22), 233 (11), 140 (21); IR (KBr) 3072, 2922, 1477, 1173, 1124, 755, 698 cm⁻¹. Anal. Calcd for $C_{17}H_{13}F_3N_2$: C, 67.54; H, 4.33. Found: C, 67.80; H, 4.40.

4.3.12. 3-Trifluoromethyl-4-(p-methoxy)phenyl-1-methyl-5-phenylpyrazole (**3***l*)

31 was prepared in 82% yield (0.027 g) according to the general procedure (the reactor was heated at 180 °C for 1.5 h). **31**: mp 98–99 °C; ¹H NMR (CDCl₃) δ 7.39–7.35 (m, 3H), 7.22–7.17 (m, 2H), 7.09–7.03 (m, 2H), 6.81–6.75 (m, 2H), 3.85 (s, 3H), 3.76 (s, 3H); ¹⁹F NMR (CDCl₃, internal standard CFCl₃) δ –59.84 (s, 3F); MS, *m*/*z* (relative intensity) 332 (*M*⁺, 100), 317 (34), 269 (12), 205 (9), 148 (10), 76 (7); IR (KBr) 3064, 2927, 1522, 1250, 1171, 1124, 817, 745, 690 cm⁻¹. Anal. Calcd for C₁₈H₁₅F₃N₂O: C, 65.06; H, 4.55. Found: C, 64.88; H, 4.51.

4.3.13. 3-Trifluoromethyl-4-(p-fluoro)phenyl-1-methyl-5-phenylpyrazole (3m)

3m was prepared in 84% yield (0.027 g) according to the general procedure (the reactor was heated at 180 °C for 1.5 h). **3m**: oil; ¹H NMR (CDCl₃) δ 7.43–6.89 (m, 9H), 3.87 (s, 3H); ¹⁹F NMR (CDCl₃, internal standard CFCl₃) δ –59.85 (s, 3F), –115.31 (s, 1F); MS, *m*/*z* (relative intensity) 320 (*M*⁺, 100), 299 (17), 251 (8), 140 (11); IR (neat) 3062, 2920, 1522, 1323, 1174, 1140, 1134, 831, 747, 695 cm⁻¹. Anal. Calcd for C₁₇H₁₂F₄N₂: C, 63.75; H, 3.78. Found: C, 63.69; H, 3.75.

4.3.14. 3-Trifluoromethyl-4-(*p*-trifluoromethyl)phenyl-1methyl-5-phenylpyrazole (**3n**)

3n was prepared in 82% yield (0.030 g) according to the general procedure (the reactor was heated at 180 °C for 1.5 h). **3n**: oil; ¹H NMR (CDCl₃) δ 7.52–7.16 (m, 9H), 3.87 (s, 3H); ¹⁹F NMR (CDCl₃, internal standard CFCl₃) δ –59.71 (s, 3F), –63.12 (s, 3F); MS, *m*/*z* (relative intensity) 370 (*M*⁺, 100), 349 (14), 301 (22), 281 (24), 165 (7), 118 (8), 77 (10); IR (neat) 3064, 2922, 1491, 1327, 1173, 1126, 809, 753, 692 cm⁻¹. Anal. Calcd for C₁₈H₁₂F₆N₂: C, 58.39; H, 3.27. Found: C, 58.51; H, 3.23.

4.3.15. 4-Allyl-3-trifluoromethyl-1-methyl-5phenylpyrazole (**30**)

30 was prepared in 90% yield (0.024 g) according to the general procedure (the reactor was heated at 180 °C for 1.5 h). **30**: oil; ¹H NMR (CDCl₃) δ 7.55–7.45 (m, 3H), 7.39–7.28 (m, 2H), 5.83 (ddt, *J* = 17.2, 9.9, 6.2 Hz, 1H), 4.97 (d, *J* = 9.9 Hz, 1H), 4.86 (d, *J* = 17.2 Hz, 1H), 3.78 (s, 3H), 3.18 (d, *J* = 6.2 Hz, 2H); ¹⁹F NMR (CDCl₃, internal standard CFCl₃) δ –61.34 (s, 3F); MS, *m/z* (relative intensity) 266 (*M*⁺, 100), 251 (71), 231 (58), 197 (22), 115 (31), 77 (23); IR (neat) 3082, 2918, 1641, 1490, 1277, 1167, 1124, 979, 911, 755, 693 cm⁻¹. Anal. Calcd for C₁₄H₁₃F₃N₂: C, 63.15; H, 4.92. Found: C, 62.89; H, 4.86.

4.3.16. 5-Cyclopropyl-3-trifluoromethyl-1,4diphenylpyrazole (**3***p*)

3p was prepared in 91% yield (0.030 g) according to the general procedure (the reactor was heated at 180 °C for 2.5 h). **3p**: mp 78–80 °C; ¹H NMR (CDCl₃) δ 7.65–7.36 (m, 10H), 1.88–1.77 (m, 1H), 0.69–0.60 (m, 2H), 0.25–0.16 (m, 2H); ¹⁹F

NMR (CDCl₃, internal standard CFCl₃) δ –59.73 (s, 3F); MS, *m/z* (relative intensity) 328 (*M*⁺, 59), 313 (29), 217 (13), 115 (26), 91 (23), 77 (100). Anal. Calcd for C₁₉H₁₅F₃N₂: C, 69.50; H, 4.60. Found: C, 69.23; H, 4.55.

4.3.17. 4-Allyl-5-cyclopropyl-3-trifluoromethyl-1phenylpyrazole (**3***q*)

3q was prepared in 93% yield (0.027 g) according to the general procedure (the reactor was heated at 180 °C for 2.5 h). **3q**: oil; ¹H NMR (CDCl₃) δ 7.56–7.36 (m, 5H), 6.04–5.85 (m, 1H), 5.08–4.98 (m, 2H), 3.42 (d, *J* = 6.0 Hz, 2H), 1.80–1.67 (m, 1H), 0.87–0.76 (m, 2H), 0.50–0.42 (m, 2H); ¹⁹F NMR (CDCl₃, internal standard CFCl₃) δ –61.21 (s, 3F); MS, *m/z* (relative intensity) 292 (*M*⁺, 13), 127 (11), 105 (13), 77 (44), 40 (100). Anal. Calcd for C₁₆H₁₅F₃N₂: C, 65.75; H, 5.17. Found: C, 65.48; H, 5.15.

4.3.18. 5-Cyclopropyl-3-trifluoromethyl-1-methyl-4phenylpyrazole (**3r**)

3r was prepared in 75% yield (0.020 g) according to the general procedure (the reactor was heated at 180 °C for 2.5 h). **3r**: mp 65–66 °C; ¹H NMR (CDCl₃) δ 7.43–7.28 (m, 5H), 3.99 (s, 3H), 1.79–1.65 (m, 1H), 0.88–0.78 (m, 2H), 0.37–0.29 (m, 2H); ¹⁹F NMR (CDCl₃, internal standard CFCl₃) δ –59.78 (s, 3F); MS, *m/z* (relative intensity) 266 (*M*⁺, 100), 251 (40), 231 (72), 152 (24), 115 (40), 76 (33). Anal. Calcd for C₁₄H₁₃F₃N₂: C, 63.15; H, 4.92. Found: C, 62.91; H, 4.84.

4.3.19. 4-Allyl-5-cyclopropyl-3-trifluoromethyl-1methylpyrazole (**3s**)

3s was prepared in 80% yield (0.019 g) according to the general procedure (the reactor was heated at 180 °C for 2.5 h). **3s**: oil; ¹H NMR (CDCl₃) δ 5.97–5.78 (m, 1H), 5.03–4.86 (m, 2H), 3.92 (s, 3H), 3.32 (d, *J* = 5.5 Hz, 2H), 1.66–1.52 (m, 1H), 1.06–0.95 (m, 2H), 0.75–0.67 (m, 2H); ¹⁹F NMR (CDCl₃, internal standard CFCl₃) δ –61.24 (s, 3F); MS, *m/z* (relative intensity) 230 (*M*⁺, 72), 189 (100), 120 (20), 69 (8). Anal. Calcd for C₁₁H₁₃F₃N₂: C, 57.39; H, 5.69. Found: C, 57.20; H, 5.72.

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